

MARK SCHEME for the May/June 2008 question paper

0620 CHEMISTRY

0620/05

Paper 5 (Practical Test), maximum raw mark 40

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All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

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- 1 Experiment 1(a) green precipitate (1) [1]
- Experiment 2(d) brown/orange/rust precipitate (1) [1]
- Table of results
- Experiment 1**
initial and final volume boxes correctly completed (1)
- Experiment 2**
initial and final volume boxes correctly completed (2)
- differences completed correctly (1)
- comparable to Supervisor's results $\pm 5 \text{ cm}^3$ (2) [6]
- (e) (pale) green (1) [1]
- (f) (turns) yellow (1) [1]
- (g) (i) Experiment 1 (1) ecf [1]
- (ii) more in Experiment 1/greater volume (1) [1]
- (iii) solution **A** more concentrated/stronger than **B** (1) $\times 2$ (1) [2]
- (h) twice the volume value for Experiment 2 (1) cm^3/ml (1) [2]
- (i) change e.g. repeat titrations (1) or use a burette/pipette
- explanation e.g. average reading more accurate (1) instead of m/cylinder [2]
- (j) (i) iron(II) present (1) [1]
- (ii) iron(III) (1) [1]

[Total: 20]

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2 Tests on solid T

(a) white (1) see Supervisor's results [1]

(b) condensation (1) solid goes black/chars (1)
 gas ignites/catches fire or flame (1) [max 2]

(c) (i) orange/yellow (1) pH6/5 (1) see Supervisor's results [2]

 (ii) white (1) precipitate (1) [2]

 (iii) reddish (1) brown on heating (1) [2]

Tests on solid V

(d) green (1) [1]

(e) solid changes colour to copper/brown (1) [1]

(f) smells acidic/vinegar or similar (1) [1]

 (i) green (1) pH 7/6 (1) see Supervisor's results [2]

 (ii) blue (1) precipitate (1) [2]

 (iii) green (1) [1]

(g) calcium present (1) [1]

(h) copper present (1) ethanoic acid/organic salt (1) [2]

[Total: 20]

[Total for paper: 40]